

B.Sc. Semester - 5 (Remedial) (CBCS) Examination
March/April-2023 (NEW COURSE)
Inorganic and Industrial Chemistry(Core)

Time: 2:30 Hours

Marks: 70

Instructions:

1. All questions are compulsory.
2. Figures to the right indicate marks.

Q. 1 (A) Calculate 18 electrons & EAN for following metal carbonyl: [04]

(1) $\text{Cr}(\text{CO})_6$, (2) $\text{Fe}(\text{CO})_5$

Q. 1 (B) Answer the following (Any Two) [10]

(1) Draw the full structure; find the number of bridge CO & non-bridge CO in

(a) $\text{Mn}_2(\text{CO})_{10}$, (b) $\text{Fe}_2(\text{CO})_9$ & (c) $\text{Fe}_3(\text{CO})_{12}$

(2) Prove that $\psi = A \sin ax + B \cos ax$ is a solution for the equation $\frac{d^2\psi}{dx^2} + \alpha^2\psi = 0$.

(3) Derive the equation of wave function and equation of energy for particle in three dimensional box.

Q. 2 (A) Answer the following: [04]

Explain Sorel's cement and Acid resistant cement.

Q. 2 (B) Answer the following (Any Two) [10]

(1) Explain manufacture of Urea by Sindri method and action of urea as a fertilizer.

(2) Explain manufacture of triple super phosphate with flow diagram.

(3) Explain burning process with equation in manufacture of Portland cement.

Q. 3 (A) Answer the following: [04]

Explain: Factor affecting on d- orbital splitting energy (Δ_o).

Q. 3 (B) Answer the following (Any Two) [10]

(1) If CFSE of $[\text{CoBr}_6]^{-3}$ is -100 KJ/mol & pairing energy is $20,000 \text{ cm}^{-1}$ then calculate splitting energy. ($1 \text{ KJ} = 83.6 \text{ cm}^{-1}$)

(2) Explain: Processes for hydrogenation of vegetable oil.

(3) Write note on: Fused silica glass and photosensitive glass.

Q. 4 (A) Answer the following: [04]

(1) Explain degeneracy with examples.

Q. 4 (B) Answer the following (Any Two) [10]

(1) Discuss: Linear & bent NO with examples.

(2) Explain: Manufacture of Ammonium nitrate by Prilling method.

(3) Explain: Mortar and Concrete.

Q. 5 (A) Answer the following. [04]

Explain: (a) Saponification value and (b) Iodine value

Q. 5 (B) Answer the following (Any Two) [10]

(1) Describe: d-orbital splitting in Oh complexes.

(2) Explain raw materials used in manufacture of glass.

(3) Explain: Orbital rotation contribution in various crystal fields.
